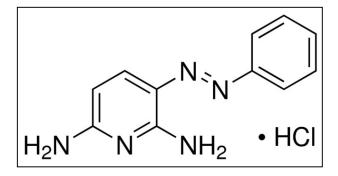
# Phenazopyridine HCl 200 mg Tablet

#### Structure:



Molecular Formula and Mass: C11H11N5 · HCl – 249.70

Category: urinary tract analgesic

#### Sample:

Grind one tablet and dissolve in 100 mL methanol. Shake at least 10 min. Concentration of solution = 200 mg/100 mL = 2.00 mg/mL. Solution is then filtered and 1.00 mL is further diluted with an additional 11.0 mL methanol. Final concentration of sample solution = 0.167 mg/mL, which is the required concentration representing 100%. **Standards:** 

High Standard:

The high limit is 115%; therefore the concentration of the high standard = (0.167 mg/mL X 1.15 = 0.192 mg/mL. Weigh approximately 19.2 mg of standard. If you weighed 19.3 mg of standard, dissolve it in: (19.3 mg)/(0.192 mg/mL) = 101 mL of methanol. This makes the high standard solution concentration equal to 0.192 mg/mL. Low Standard:

The low limit is 85%; therefore the concentration of the low standard = (0.167 mg/mL X 0.85 = 0.142 mg/mL. Dilute 1.00 mL of high standard to 1.35 mL by adding 0.35 mL of methanol (1.15/0.85 = 1.35).

## Spotting:

Spot on the 5 X 10 cm silica gel TLC aluminium plate with 3 μL aliquots as follows: Left spot low standard (85%) = 0.426 μg Center Spot 100% sample = 0.501 μg

Right Spot high standard (115%) =  $0.576 \,\mu g$ 

## **Development:**

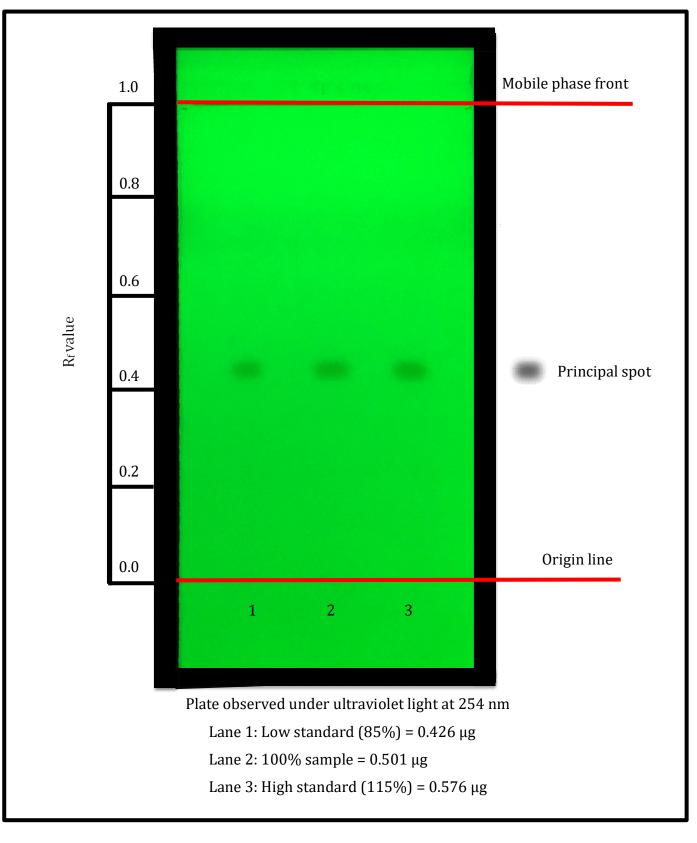
Mix 18.0 mL of ethyl acetate, 4.00 mL acetone, and 0.100 mL glacial acetic acid. Develop the plate in a small glass chamber with approximately 20.0 mL of this solution until the solvent front reaches within 1 cm of the top of the TLC plate.

 $(R_f = 0.42)$ 

## **Detection:**

<u>UV:</u>

Dry the plate and observe under ultraviolet light at 254 nm. Observe the intensities and the sizes of the spots.



Developed and tested by Kaitlin Nguyen and Joseph Sherma Department of Chemistry, Lafayette College, Easton, PA, USA March 2017